

Synthesis of *N'*-(2,3-dihydroxybenzylidene)-4-methylbenzohydrazide: Spectroscopic Characterization and X-ray Structure Determination

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To cite this article:

Alioune Fall, Mohamedou El Boukhary, Thierno Mousa Seck, Farba Bouyagui Tamboura, Ibrahima Elhadj Thiam, Aliou Hamady Barry, Mohamed Gaye, Javier Ellena. Synthesis of *N'*-(2,3-dihydroxybenzylidene)-4-methylbenzohydrazide: Spectroscopic Characterization and X-ray Structure Determination. *Modern Chemistry*. Vol. 11, No. 2, 2023, pp. 43-48. doi: 10.11648/j.mc.20231102.11

Received: April 11, 2023; Accepted: April 27, 2023; Published: May 10, 2023

Abstract: p-Toluic hydrazide was used for synthesizing a new Schiff base *N'*-(2,3-dihydroxybenzylidene)-4-methylbenzohydrazide (I). The prepared compound was characterized by elemental analysis, infrared and ¹H and ¹³C NMR spectroscopy techniques, and the structure of compound (I) was determined by single-crystal X-ray diffraction study. The compound (C₁₅H₁₄N₂O₃) crystallises in the monoclinic space group *P*2₁/*n* with the following unit cell parameters: *a* = 11.7820 (6) Å, *b* = 8.5278 (4) Å, *c* = 13.6054 (7) Å, *β* = 109.346 (5)°, *V* = 1289.81 (12) Å³, *Z* = 4, *R*₁ = 0.071 and *wR*₂ = 0.197. The X-ray structure shows that the compound (I) adopts an *E* configuration with respect to the double bond C7=N1. The carbonyl oxygen atom O3 and the nitrogen atom N1 of the hydrazide moiety are in a *syn* conformation with respect to C8—N2 bond. The crystal packing of compound (I) is stabilized by intramolecular O(phenol)—H···N(carbohydrazide) which results in an *S*(6) ring motif and intermolecular O(phenol)—H···O(carbohydrazide) hydrogen bonds which form chains. The carbonohydrazide moiety C=N—N—C(O)—N—N=C fragment is almost coplanar with the two benzene rings with dihedral angles of their mean planes of 10.35° [C1—C6] and 7.99° [C9—C14], respectively. The dihedral angle between the mean planes of the phenyl rings is 2.46°.

Keywords: Schiff Base, 2,3-dihydroxybenzaldehyde, p-toluic Hydrazide, Single Crystal Structure

1. Introduction

The hydrazides are compounds possessing a very reactive arm with carbonyl compounds. Control of the hydrazide/carbonyl ratio allows to synthesized mono, di or tri condensation compounds. In these types of compounds, the part containing the azomethine group connected to a carbonyl group (-HC=N-NH-C(O)-) is responsible for certain pharmaceutical properties [1–4]. Hydrazide derivatives have been used as precursors for the synthesis of various heterocyclic compounds containing a nitrogen atom in the ring and/or in the free arms [5–7]. The reactivity as well as the

biological properties of compounds derived from hydrazide have been widely studied for the development of drugs or phytosanitary products [8–11]. They are known to possess a wide spectrum of biological activities such as antioxidant [12], analgesic [13], antifungal [14], antimicrobial [15], anti-spasmodic [16], anti-inflammatory [17], antituberculosis [18], and anticancer [19, 20]. These Schiff bases are also known for their physical properties. They are used in the field of fluorescence [21, 22], optics [23] and corrosion inhibition [24]. In the field of coordination chemistry, these Schiff bases are recognized as multitopic ligands for the targeted construction of original

metallo-organic architectures [25–27]. In this context, we report the synthesis and crystal structure of a derivative of *p*-toluic acid hydrazide and 2,3-dihydroxybenzaldehyde.

2. Experimental

2.1. Starting Materials and Instrumentations

p-toluic hydrazide and 2,3-dihydroxybenzaldehyde were purchased from Sigma-Aldrich and used as received without further purification. All solvents used were of reagent grade. Melting points were determined on a Büchi 570 melting-point apparatus and were uncorrected. Elemental analyses of C, H and N were recorded on a VxRio EL Instrument. Infrared spectra were obtained on a FTIR Spectrum Two of Perkin Elmer spectrometer in the 4000–400 cm^{-1} region. The ^1H and ^{13}C NMR spectra were recorded in DMSO- d_6 on a BRUKER 300.13 MHz spectrometer at room temperature using TMS as an internal reference.

2.2. Synthesis of

N'-(2,3-dihydroxybenzylidene)-4-methylbenzohydrazide (H_3L) (I)

p-toluic hydrazide (1.5 g, 10 mmol) and 2,3-dihydroxybenzaldehyde (1.38 g, 10 mmol) were mixed in ethanol and one drop of glacial acetic acid was added. The mixture was stirring under reflux for 120 minutes. On cooling the white precipitate which appears was recovered by filtration. The precipitate was washed with ethanol before being dried in air. Recrystallization of the compound in ethanol afforded white crystal suitable for X-ray analyze after two days. Yield 85%. M.p. > 260 °C. Anal. Calc. for $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_3$: C, 66.66; H, 5.22; N, 10.36%. Found: C, 66.64; H, 5.20; N, 10.33%. FTIR (ν , cm^{-1}): 3287 (N—H); 1684 (C=O); 1611 (C=N); 1557–1481 $\text{C}_{\text{Ar}}=\text{C}_{\text{Ar}}$; 1266 (C—O) $_{\text{phenol}}$; 1070 (N—N). ^1H NMR (DMSO- d_6 , δ (ppm)): 2.5 (3H, s, CH₃), 6.8 (3H, mult, H—Ar), 7.4 (2H, D, H—Ar), 7.8 (2H, D, H—Ar), 8.6 (1H, S, H—C=N), 9.3 (1H, S, H—N), 11.3 (1H S, H—OPh), 12.2 (1H S, H—OPh). ^{13}C NMR (DMSO- d_6 , δ (ppm)): 163.10 (C=O); 149.35 (C=N); 146.60 ($\text{C}_{\text{Ar}}-\text{O}$), 146.08 ($\text{C}_{\text{Ar}}-\text{O}$), 142.5 (C_{Ar}), 130.54 (C_{Ar}), 129.55 (C_{Ar}), 128.13 (C_{Ar}), 120.64 (C_{Ar}), 119.60 (C_{Ar}), 119.22 (C_{Ar}), 117.84 (C_{Ar}), 21.46 (—CH₃).

2.3. Crystal Structure Determination

Crystals suitable for single-crystal X-ray diffraction, of the reported compound, was grown by slow evaporation of EtOH solution of the compound. Details of the crystal structure solution and refinement are given in Table 1. Diffraction data were collected using a Bruker APEX-II CCD diffractometer with graphite monochromatized MoK α radiation ($\lambda = 0.71073$ Å). All data were corrected for Lorentz and polarization effects. The structure was solved and refined using the Bruker SHELXTL Software Package [28]. All the structures were refined on F^2 by a full-matrix least-squares procedure using anisotropic displacement parameters for all non-hydrogen atoms [29]. H atoms of the NH group was located in the

Fourier difference maps and refined without restraints. Other H atoms were geometrically optimized and refined as riding on their carriers with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ (1.5 for CH₃ groups). Molecular graphics were generated using ORTEP-3 [30].

Table 1. Crystallographic data and refinement parameters for compound H_3L (I).

Chemical formula	$\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_3$
M_r	270.28
Crystal shape/color	Prismatic/Dark yellow
Crystal size (mm)	0.13 × 0.11 × 0.06
Crystal system, space group	Monoclinic, $P2_1/n$
a (Å)	11.7820 (6)
b (Å)	8.5278 (4)
c (Å)	13.6054 (7)
β (°)	109.346 (6)
V (Å ³)	1289.81 (12)
Z	4
D_{cal} (g cm ⁻³)	1.392
Temperature (K)	100
Radiation type	Mo K α
μ (mm ⁻¹)	0.10
Index ranges	-14 ≤ h ≤ 14; -10 ≤ k ≤ 10; -16 ≤ l ≤ 16
$F(000)$	568
q range (°)	2.851–34.063
No. of measured reflections	15026
No. of independent reflections	2449
No. of observed [$I > 2\sigma(I)$] reflections	2983
R_{int}	0.055
$R[F^2 > 2\sigma(F^2)]$	0.056
$wR(F^2)$	0.160
Goodness-of-fit (GOF)	1.07
No. of parameters	184
No. of restraints	0
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å ⁻³)	0.36, -0.28

3. Result and Discussion

3.1. General Study

The compound obtained after the condensation reaction between *p*-toluic hydrazide and 2,3-dihydroxybenzaldehyde (Figure 1) is characterized by physico-chemical and spectroscopic analyses. Elemental analysis of the compound confirms the formation of the compound with the chemical formula $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_3$. The FTIR spectrum of the H_3L (1) ligand shows bands pointed at 3286 and 3140 cm^{-1} attributable respectively to the phenolic ν_{OH} and ν_{NH} vibrations of the hydrazide function [31]. The vibration of the $\nu_{\text{C=O}}$ carbonyl appears at 1634 cm^{-1} while the $\nu_{\text{C=N}}$ band is pointed at 1611 cm^{-1} [32]. Between 1557 and 1481 cm^{-1} we identify the bands corresponding to the $\nu_{\text{C=C}}$ vibrations of the aromatic ring. The FTIR spectrum also reveals a middle band at 1242 cm^{-1} attributed to the $\nu_{\text{C=N}}$ vibration and a strong band at 1266 cm^{-1} attributed to the phenolic $\nu_{\text{C-O}}$ bonding vibration. The band that appears at 1070 cm^{-1} is due to the $\nu_{\text{N-N}}$ vibration. The proton NMR spectrum confirms the proposed formula. Indeed, two characteristic phenolic proton signals are pointed at 12.2 and 11.3 ppm. A signal at 8.6 ppm attributed to the proton of the azomethine group confirms the condensation. At 9.3 ppm, a signal characteristic of the presence of HN— appears. The

signals that appear in the 6.8-7.8 ppm region are due to the protons of the aromatic nuclei. The protons of the CH₃-Ar group appear around 2.5 ppm as expected for protons of this type. The carbon NMR spectrum shows characteristic signals from the carbon atoms of the C=O and C=N groups at 163.10 and 149.35 ppm, respectively. The C_{ipso}-OH carbon atoms appear at 146.60 and 146.08 ppm, respectively.

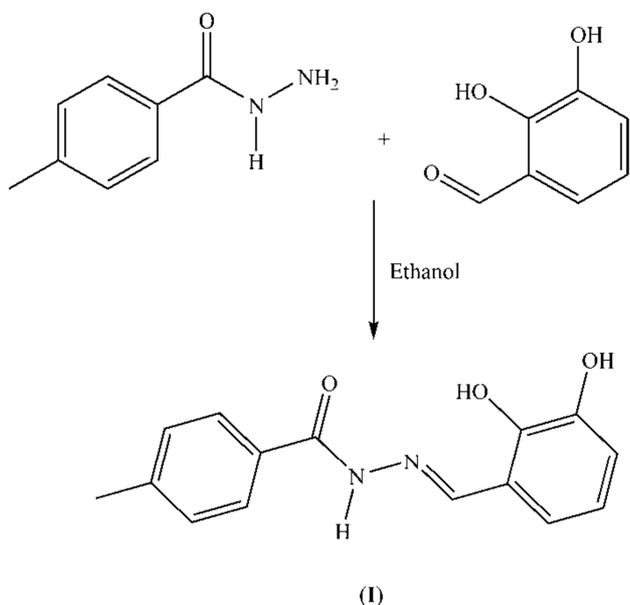


Figure 1. Synthetic scheme for the ligand (I).

3.2. X-ray Structure

The compound (I) crystallizes in Monoclinic, $P2_1/n$. The asymmetric unit contains one ligand molecule of the Schiff base ligand. The molecular structure of the title compound (I) with the atomic-labeling scheme is shown in Figure 2. Selected bond distances and torsion angle are listed in Table 2. The Schiff base adopts an *E* configuration with respect to the

C7=N1. The carbonyl oxygen atom O3 and the nitrogen atom N1 of the hydrazide moiety are in a *syn* conformation with respect to C8—N2 [O3—C8—N2—N1 = 2.1 (3)°]. The carbonylhydrazide C7=N1—N2—C8=O3 moiety is almost coplanar with the two benzene rings with dihedral angles of their mean planes of 10.35° [C1—C6] and 7.99° [C9—C14]. The two benzene rings are almost coplanar with dihedral angle of 2.46°. The C8=O3 and C7=N1 bond length of 1.240 (2) Å and 1.283 (3) Å, respectively, which are double-character are indicative of non-enolization of the compound (I) which exist only in its amid form. The C8—N2 [1.348 (3) Å] and the N1—N2 [1.379 (2) Å] distances indicate that these correspond to single bonds and are comparable to those observed for 1-(1-Phenylethylidene)carbohydrazide [33]. The intramolecular hydrogen bonds O1(phenol)—H···N1(azomethine) forms an *S*(6) ring motif. In the crystal, each independent molecule (I) donates hydrogen bonds to the carbonyl oxygen atom of one ligand molecule, forming a chain [O2(phenol)—H···O3ⁱ(carbonyl)]; $i = -x+3/2, y+1/2, -z+3/2$] (Figure 3, Table 3).

Table 2. Selected bond lengths (Å) and torsion angles (deg) for the compound (I).

Atom-Atom		Atom-Atom-Atom-Atom	
O3—C8	1.240 (2)	C14—C9—C8—O3	9.6 (3)
O1—C1	1.352 (2)	C8—N2—N1—C7	170.1 (2)
O2—C2	1.369 (2)	N1—C7—C6—C1	-2.5 (3)
N2—N1	1.379 (2)	C10—C9—C8—O3	172.1 (2)
N2—C8	1.348 (3)	O3—C8—N2—N1	2.1 (3)
N1—C7	1.283 (3)	N2—N1—C7—C6	-179.4 (2)

Table 3. Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H1···N1	0.84	1.90	2.630 (2)	144.6
O2—H2···O3 ⁱ	0.84	1.83	2.6647 (19)	171.5
N2—H2A···O2 ⁱⁱ	0.88	2.19	2.914 (2)	139.6

Symmetry codes: (i) $-x+3/2, y+1/2, -z+3/2$; (ii) $x-1/2, -y+3/2, z-1/2$.

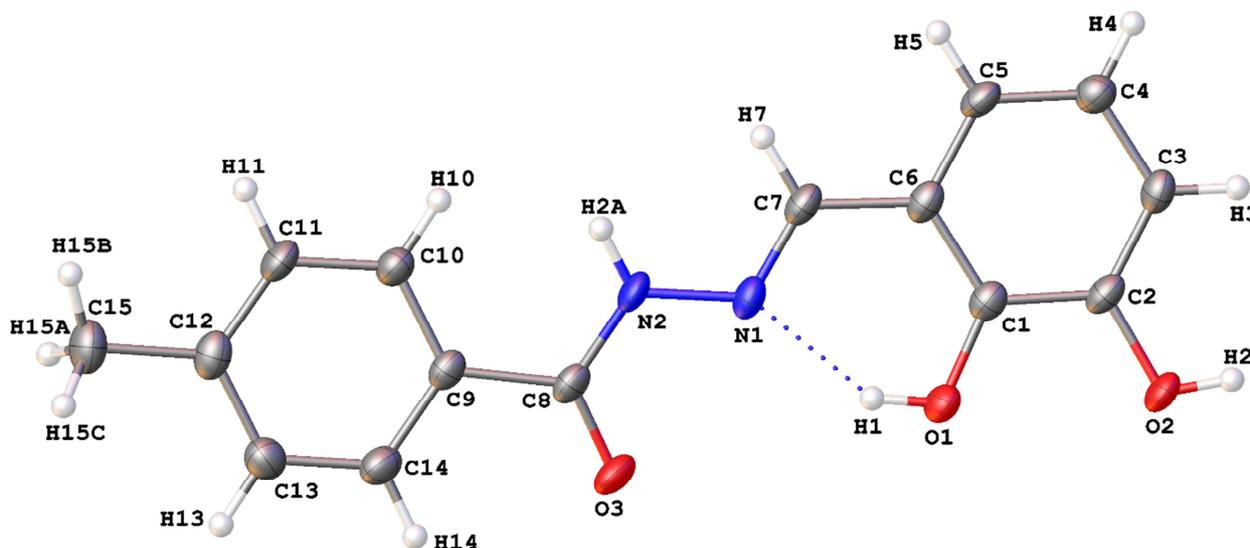


Figure 2. ORTEP plot (30% probability ellipsoids) showing the structure of (I).

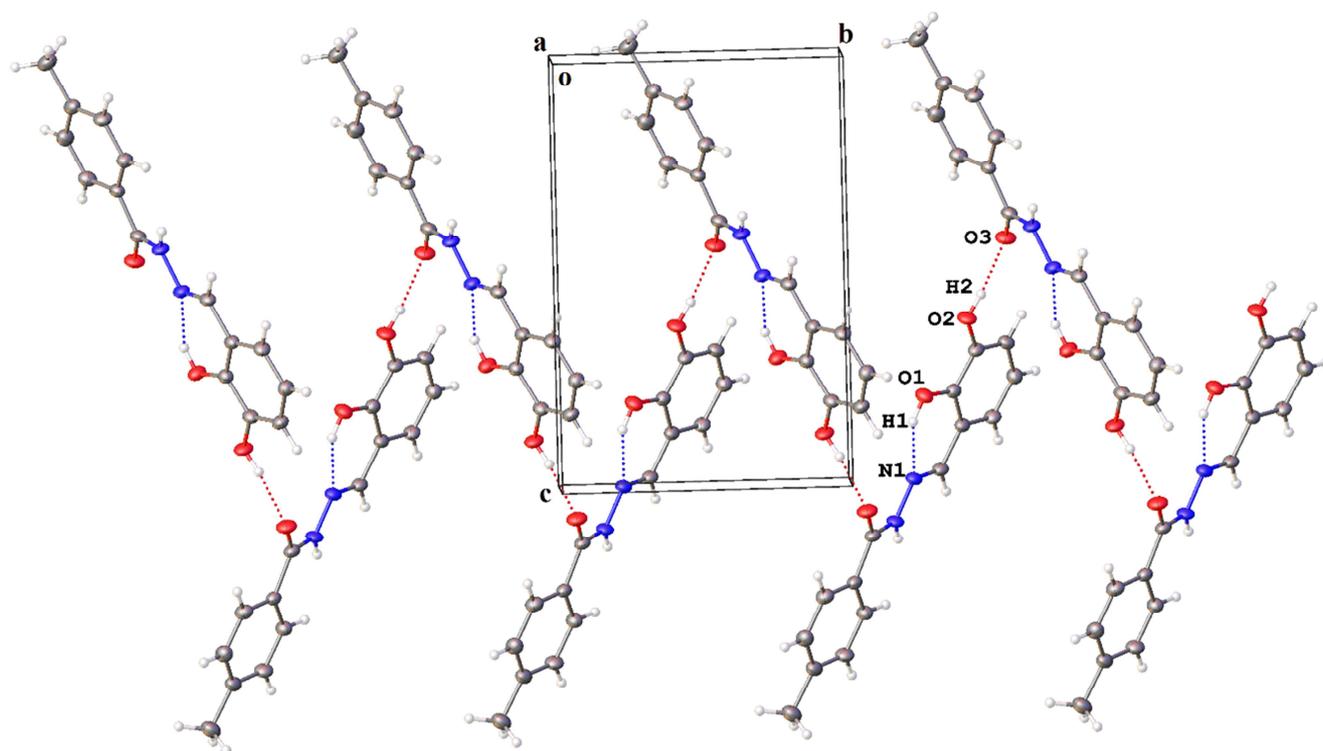


Figure 3. Chain of the ligand formed by hydrogen bonding.

4. Conclusion

The hydrazone derivative *N'*-(2,3-dihydroxybenzylidene)-4-methylbenzohydrazide (I) was prepared from *p*-toluic acid hydrazide and 2,3-dihydroxybenzaldehyde. The structure of (I) was confirmed by elemental analysis and spectroscopic techniques (FT-IR, ^1H and ^{13}C NMR). The molecular structure of (I) was determined by X-ray diffraction technique. The X-ray structure reveals that the carbohydrazide moiety and the two benzene ring are almost coplanar. This compound present several donor sites such as nitrogen and oxygen atoms. It can be used as Schiff base ligand for the synthesis of coordination compounds.

Supporting Information

CCDC-2253533 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via <https://www.ccdc.cam.ac.uk/structures/>.

References

- [1] Rollas, S. & Küçükgül, S. G. (2007). Biological Activities of Hydrazone Derivatives. *Molecules*, 12 (8), 1910–1939. <https://doi.org/10.3390/12081910>
- [2] Narang, R., Narasimhan, B. & Sharma, S. A Review on Biological Activities and Chemical Synthesis of Hydrazone Derivatives. *Current Medicinal Chemistry*, 19 (4), 569–612. <https://doi.org/10.2174/092986712798918789>
- [3] Mohareb, R. M., Fleita, D. H. & Sakka, O. K. (2011). Novel Synthesis of Hydrazone-Hydrazone Derivatives and Their Utilization in the Synthesis of Coumarin, Pyridine, Thiazole and Thiophene Derivatives with Antitumor Activity. *Molecules*, 16 (1), 16–27. <https://doi.org/10.3390/molecules16010016>
- [4] Popiołek, Ł. & Biernasiuk, A. (2016). Design, synthesis, and in vitro antimicrobial activity of hydrazone-hydrazone of 2-substituted acetic acid. *Chemical Biology & Drug Design*, 88 (6), 873–883. <https://doi.org/10.1111/cbdd.12820>
- [5] Zala, M., Vora, J. J. & Patel, H. B. (2020). Synthesis, Characterization, and Comparative Study of Some Heterocyclic Compounds Containing Isoniazid and Nicotinic Acid Hydrazone Moieties. *Russian Journal of Organic Chemistry*, 56 (10), 1795–1800. <https://doi.org/10.1134/S1070428020100218>
- [6] Saleh, A. & Saleh, M. Y. (2022). Synthesis of heterocyclic compounds by cyclization of Schiff bases prepared from capric acid hydrazide and study of biological activity. *Egyptian Journal of Chemistry*, 65 (12), 783–792. <https://doi.org/10.21608/EJCHEM.2022.133946.5904>
- [7] Sharma, B., Chowdhary, S., Legac, J., Rosenthal, P. J. & Kumar, V. (2023). Quinoline-based heterocyclic hydrazones: Design, synthesis, anti-plasmodial assessment, and mechanistic insights. *Chemical Biology & Drug Design*, 101 (4), 829–836. <https://doi.org/10.1111/cbdd.14185>
- [8] Popiołek, Ł. (2017). Hydrazone-hydrazone as potential antimicrobial agents: overview of the literature since 2010. *Medicinal Chemistry Research*, 26 (2), 287–301. <https://doi.org/10.1007/s00044-016-1756-y>
- [9] Popiołek, Ł. (2021). Updated Information on Antimicrobial Activity of Hydrazone-Hydrazone. *International Journal of Molecular Sciences*, 22 (17), 1795–1800. <https://doi.org/10.3390/ijms22179389>

- [10] Bedia, K.-K., Elçin, O., Seda, U., Fatma, K., Nathaly, S., Sevim, R. & Dimoglo, A. (2006). Synthesis and characterization of novel hydrazide-hydrazones and the study of their structure-antituberculosis activity. *European Journal of Medicinal Chemistry*, 41 (11), 1253–1261. <https://doi.org/10.1016/j.ejmech.2006.06.009>
- [11] Turan-Zitouni, G., Altıntop, M. D., Özdemir, A., Demirci, F., Mohsen, U. A. & Kaplancıklı, Z. A. (2013). Synthesis and antifungal activity of new hydrazide derivatives. *Journal of Enzyme Inhibition and Medicinal Chemistry*, 28 (6), 1211–1216. <https://doi.org/10.3109/14756366.2012.723208>
- [12] Aslanhan, Ö., Kalay, E., Tokalı, F. S., Can, Z. & Şahin, E. (2023). Design, synthesis, antioxidant, and anticholinesterase activities of novel isonicotinic hydrazide-hydrazone derivatives. *Journal of Molecular Structure*, 1279, 135037. <https://doi.org/10.1016/j.molstruc.2023.135037>
- [13] Nocheva, H., Vladimirova, S., Tzankova, D., Peikova, L. & Georgieva, M. (2021). Analgesic properties of newly synthesized N pyrrolyl hydrazide hydrazones. *Tropical Journal of Pharmaceutical Research*, 22 (1), 121–127. <https://doi.org/10.4314/tjpr.v22i1.17>
- [14] Mallikarjuna, B. P., Sastry, B. S., Kumar, G. V. S., Rajendraprasad, Y., Chandrashekar, S. M. & Sathisha, K. (2009). Synthesis of new 4-isopropylthiazole hydrazide analogs and some derived clubbed triazole, oxadiazole ring systems – A novel class of potential antibacterial, antifungal and antitubercular agents. *European Journal of Medicinal Chemistry*, 44 (11), 4739–4746. <https://doi.org/10.1016/j.ejmech.2009.06.008>
- [15] Sethiya, A., Joshi, D., Manhas, A., Sahiba, N., Agarwal, D. K., Jha, P. C. & Agarwal, S. (2023). Glycerol based carbon sulfonic acid catalyzed synthesis, in silico studies and in vitro biological evaluation of isonicotinohydrazide derivatives as potent antimicrobial and anti-tubercular agents. *Heliyon*, 9 (2), e13226. <https://doi.org/10.1016/j.heliyon.2023.e13226>
- [16] Berillo, D. A. & Dyusebaeva, M. A. (2022). Synthesis of hydrazides of heterocyclic amines and their antimicrobial and spasmolytic activity. *Saudi Pharmaceutical Journal*, 30 (7), 1036–1043. <https://doi.org/10.1016/j.jsps.2022.04.009>
- [17] Verma, S., Lal, S., Narang, R. & Sudhakar, K. (2023). Quinoline Hydrazide/Hydrazone Derivatives: Recent Insights on Antibacterial Activity and Mechanism of Action. *ChemMedChem*, 18 (5), e202200571. <https://doi.org/10.1002/cmde.202200571>
- [18] Kumar, B. V. S., Khetmalis, Y. M., Nandikolla, A., Kumar, B. K., Van Calster, K., Murugesan, S., Cappelletti, D., Sekhar, K. V. G. C & Sekhar, K. V. G. C. (2023). Design, Synthesis, and Antimycobacterial Evaluation of Novel Tetrahydroisoquinoline Hydrazide Analogs. *Chemistry & Biodiversity*, 20 (2), e202200939. <https://doi.org/10.1002/cbdv.202200939>
- [19] Alam, M. S. & Lee, D.-U. Hydrazide-hydrazones as Small Molecule Tropomyosin Receptor Kinase A (TRKA) Inhibitors: Synthesis, Anticancer Activities, In silico ADME and Molecular Docking Studies. *Medicinal Chemistry*, 19 (1), 47–63. <https://doi.org/10.2174/1573406418666220427105041>
- [20] Bora, D., Sharma, A., John, S. E. & Shankaraiah, N. (2023). Development of hydrazide hydrazone-tethered combretastatin-oxindole derivatives as antimitotic agents. *Journal of Molecular Structure*, 1275, 134675. <https://doi.org/10.1016/j.molstruc.2022.134675>
- [21] He, G., Hua, X., Yang, N., Li, L., Xu, J., Yang, L., Wang, Q. & Ji, L. (2019). Synthesis and application of a “turn on” fluorescent probe for glutathione based on a copper complex of coumarin hydrazide Schiff base derivative. *Bioorganic Chemistry*, 91, 103176. <https://doi.org/10.1016/j.bioorg.2019.103176>
- [22] Yang, G., Li, P., Han, Y., Tang, L., Liu, Y., Xin, H., Wang, K.-N., Zhao, S., Liu, Z. & Cao, D. (2023). A coumarin hydrazide Schiff base fluorescent probe for sensitively sensing Al³⁺ in living cells. *Materials Chemistry and Physics*, 295, 127145. <https://doi.org/10.1016/j.matchemphys.2022.127145>
- [23] Shankar, M., Raj, A. D., Purusothaman, R., Vimalan, M., Athimoolam, S. & Potheher, I. V. (2019). Studies on optical, electrical, mechanical, and theoretical investigation of 4-nitro-benzoic acid (3-ethoxy-2-hydroxy-benzylidene)-hydrazide: A novel Schiff base organic NLO material. *Journal of Molecular Structure*, 1181, 348–359. <https://doi.org/10.1016/j.molstruc.2018.12.082>
- [24] Gupta, S. R., Mourya, P., Singh, M. M. & Singh, V. P. (2014). Synthesis, structural, electrochemical and corrosion inhibition properties of two new ferrocene Schiff bases derived from hydrazides. *Journal of Organometallic Chemistry*, 767, 136–143. <https://doi.org/10.1016/j.jorganchem.2014.05.038>
- [25] Lu, Y., Huang, Z.-M., Zou, H.-H. & Liang, F.-P. (2023). Structures and magnetic properties of two series of Schiff base binuclear lanthanide complexes. *Applied Organometallic Chemistry*, 37 (1), e6934. <https://doi.org/10.1002/aoc.6934>
- [26] Dongare, G. & Aswar, A. (2023). Synthesis of new heterocyclic N'-(2-hydroxy-3-methoxybenzylidene)-4-oxopiperidine-1-carboxamide and its mononuclear metal (II) complexes: Spectroscopic characterization, fluorescence, DFT, thermo-kinetic, and antimicrobial studies. *Journal of Molecular Structure*, 1281, 135107. <https://doi.org/10.1016/j.molstruc.2023.135107>
- [27] Seck, T. M., Gaye, P. A., Ndoye, C., Thiam, I. E., Diouf, O., Retailleau, P. & Gaye, M. (2020). Diaqua-bis-μ-1,5-bis[(pyridin-2-yl)methylidene]carbonohydrazide(1–)di-μ-chlorido-tetrachloridotetrazinc(II). *Acta Crystallographica Section E*, 76 (8), 1349–1352. <https://doi.org/10.1107/S2056989020009834>
- [28] Sheldrick, G. M. (2015). SHELXT – Integrated space-group and crystal-structure determination. *Acta Crystallographica Section A*, 71 (1), 3–8. <https://doi.org/10.1107/S2053273314026370>
- [29] Sheldrick, G. M. (2015). Crystal structure refinement with SHELXL. *Acta Crystallographica Section C*, 71 (1), 3–8. <https://doi.org/10.1107/S2053229614024218>
- [30] Farrugia, L. J. (2012). WinGX and ORTEP for Windows: an update. *Journal of Applied Crystallography*, 45 (4), 849–854. <https://doi.org/10.1107/S0021889812029111>
- [31] Gaye, M., Tamboura, F. B. & Sall, A. S. (2003). Spectroscopic studies of some lanthanide(III) nitrate complexes synthesized from a new ligand 2,6-bis-(salicylaldehyde hydrazone)-4-chlorophenol. *Bulletin of the Chemical Society of Ethiopia*, 17 (1). <https://doi.org/10.4314/bcse.v17i1.61726>
- [32] Tamboura, F. B., Diouf, O., Barry, A. H., Gaye, M. & Sall, A. S. (2012). Dinuclear lanthanide(III) complexes with large-bite Schiff bases derived from 2,6-diformyl-4-chlorophenol and hydrazides: Synthesis, structural characterization, and spectroscopic studies. *Polyhedron*, 43 (1), 97–103. <https://doi.org/10.1016/j.poly.2012.06.025>

- [33] Qiao, Y., Ju, X., Gao, Z. & Kong, L. (2010).
1-(1-Phenylethylidene)carbonohydrazide. *Acta
Crystallographica Section E*, 66 (10), o2691.
<https://doi.org/10.1107/S1600536810038353>